

## Chemical transformations of oxyresveratrol (*trans*-2,4,3',5'-tetrahydroxystilbene) into a potent tyrosinase inhibitor and a strong cytotoxic agent

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**Abstract**—From oxyresveratrol (*trans*-2,4,3',5'-tetrahydroxystilbene **1**), seven derivatives were prepared, including *trans*-2-methoxy-4,3',5'-trihydroxystilbene (**2**), *trans*-2,3'-dimethoxy-4,5'-dihydroxystilbene (**3**), *trans*-4,3'-dimethoxy-2,5'-dihydroxystilbene (**4**), *trans*-2,4,3',5'-tetramethoxystilbene (**5**) and *cis*-2,4,3',5'-tetramethoxystilbene (**6**), 2,4,3',5'-tetrahydroxybibenzyl (**7**), and 2,4,3',5'-tetramethoxybibenzyl (**8**). The tetrahydroxybibenzyl **7**, a hydrogenation product of **1**, exhibited more potent tyrosinase inhibitory activity than the parent compound, without cytotoxicity. A kinetic study revealed that **7** was a reversible and non-competitive inhibitor of mushroom tyrosinase with L-dopa as the substrate. Analysis of the  $K_i$  values indicated that **7** has a slightly higher affinity to the enzyme than **1**. Compound **6**, a tetra-*O*-methylated analogue of **1** with *cis*-configuration, was deprived of inhibitory effect on the enzyme tyrosinase, but showed very strong cytotoxicity against the human cancer cells KB, BC, and NCI-H187, with potency comparable to those of the anticancer agents ellipticine and doxorubicin. Data on the tyrosinase inhibitory activity and cytotoxicity of **1**–**8** indicated that *O* methylation on stilbene **1** destroyed anti-tyrosinase activity but generated cytotoxicity. Thus, facile preparations of a potent tyrosinase inhibitor (**7**) and a strong cytotoxic agent (**6**) from the natural product **1** were achieved through simple chemical reactions.

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Oxyresveratrol (*trans*-2,4,3',5'-tetrahydroxystilbene, **1**) is a major constituent of the heartwood of *Artocarpus lakoocha* Roxb.<sup>1,2</sup> The compound is the active principle of 'Puag-Haad,' a dried aqueous extract traditionally used in Thailand as an anthelmintic.<sup>3</sup> Recent studies have revealed that **1** also possesses potent inhibitory activity against tyrosinase,<sup>4,5</sup> a key enzyme in the biosynthetic pathway of melanin pigments in both plants and animals. Further investigations have shown that the compound has strong skin depigmenting effects in both animals and humans.<sup>6</sup> As a strong tyrosinase inhibitor, **1** has potential applications as a skin-whitening agent in cosmetic preparations or as an anti-browning agent for food products of plant origin. Other interesting biological activities reported for **1** include antiherpetic and anti-HIV,<sup>2</sup> antiinflammatory,<sup>7</sup> antioxidative,<sup>8</sup> and antiapoptotic and neuroprotective activi-

ties.<sup>9</sup> Since **1** can be obtained in large amounts from the widely available *A. lakoocha*,<sup>2–4</sup> the compound appears to be an attractive target for chemical studies in terms of structure–activity relationships. In the present communication, we describe our efforts to prepare, from **1**, analogues with higher potency and selectivity with regard to their cytotoxic and anti-tyrosinase activities.

The reaction of **1** with CH<sub>3</sub>I (3 equiv, reflux, 24 h) gave three partially methylated products, namely *trans*-2-methoxy-4,3',5'-trihydroxystilbene (**2**), *trans*-2,3'-dimethoxy-4,5'-dihydroxystilbene (**3**), and *trans*-4,3'-dimethoxy-2,5'-dihydroxystilbene (**4**). Full *O*-methylation of **1** was achieved when CH<sub>3</sub>I (6 equiv) was employed, leading to the formation of *trans*-2,4,3',5'-tetramethoxystilbene (**5**). Attempts to prepare other mono-, di-, and tri-methoxy derivatives of **1** using CH<sub>3</sub>I or (CH<sub>3</sub>)<sub>2</sub>SO<sub>4</sub> in various conditions were not successful. When *trans*-tetramethoxystilbene **5** was subjected to a photochemical reaction, the *cis*-isomer **6** was formed. Catalytic Pd/C hydrogenation of **1** and **5** gave bibenzyl compounds **7** and **8**, respectively. The positions of the

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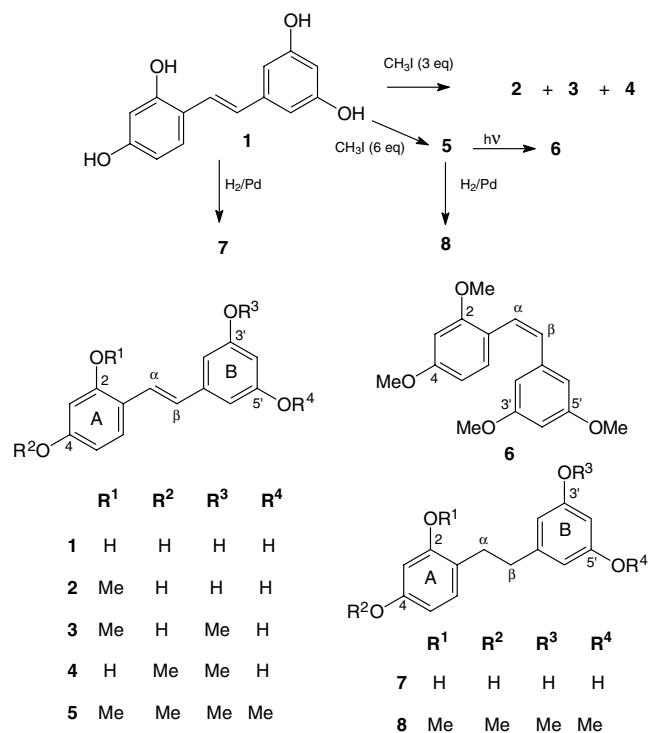


Figure 1. Chemical transformations of **1** into **2–8**.

methoxy groups in **2–4** were determined by NOESY experiments. The *cis*-configuration of **6** was established from the coupling constant between H- $\alpha$  and H- $\beta$  ( $J = 12.1$  Hz) ( $J_{\alpha\beta} = 16.5$  Hz in **5**), and from the NOESY effect observed between H-6 and H-2'(6'). These natural and synthetic stilbenoids (**1–8**) (Fig. 1) were then evaluated for tyrosinase inhibitory activity and cytotoxic potential.<sup>10</sup>

For the evaluation of tyrosinase inhibitory activity, assays were performed with L-dopa as the substrate as previously described, using kojic acid, a well-known strong tyrosinase inhibitor, as the positive control.<sup>4,11–13</sup> Table 1 summarizes the percentages of inhibition and the IC<sub>50</sub> values of **1–8**, as compared with kojic acid. It should be mentioned that oxyresveratrol (**1**) was earlier reported to be a stronger tyrosinase inhibitor than kojic acid,<sup>5</sup> and this was confirmed in this study, as reflected in their respective IC<sub>50</sub> values of 12.7 and 133.4  $\mu$ M (Table 1). For the *O*-methylated products (**2–6** and **8**), none of them showed significant inhibitory activity

(IC<sub>50</sub> > 100  $\mu$ M). The loss of activity in **2–6** and **8** was caused by the disappearance of the 4-alkyl resorcinol-like structure of ring A.<sup>12–14</sup> Compound **7**, a bibenzyl derivative prepared from **1**, showed more inhibitory activity than did the parent compound, being about 8-fold stronger in view of the IC<sub>50</sub> value. Further analysis of the data obtained for **7** indicated that this compound, as well as **1** and kojic acid, inhibited the enzyme in a dose-dependent manner (Fig. 2). The higher tyrosinase inhibitory activity of **7**, as compared with **1**, was probably due to its bibenzyl structure which gave more flexibility and thus allowed the phenolic groups to interact with the enzyme more effectively. Kinetic studies were then conducted on **7**, in comparison with **1**, with regard to the ability to inhibit mushroom tyrosinase with L-dopa as the substrate. As summarized in Table 2, the  $V_{\max}$  ( $\Delta A_{490}/\text{min}$ ) for the dopa oxidase activity of the enzyme was  $2.0 \times 10^{-1}$ , and the  $K_m$  value was 0.7 mM L-dopa. It can be seen from the Lineweaver–Burk plot of  $1/V$

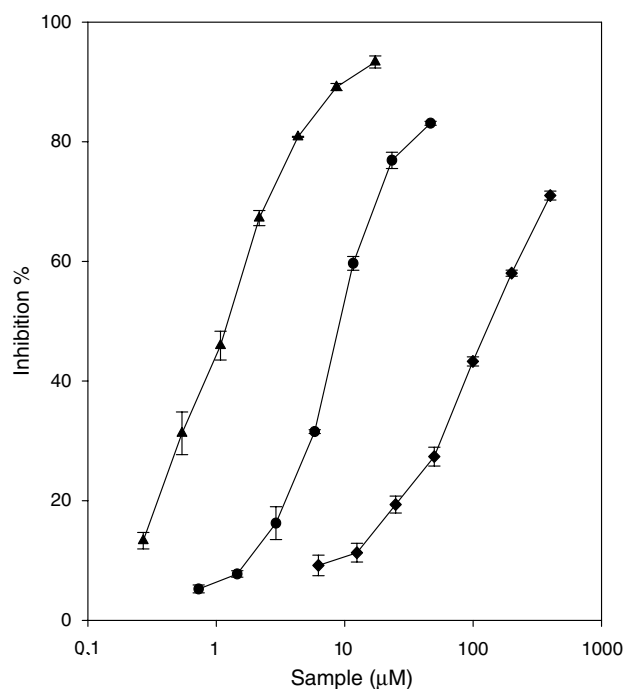


Figure 2. Dose-dependent inhibitory effects on mushroom tyrosinase by oxyresveratrol (**1**), 2,4,3',5'-tetrahydroxybibenzyl (**7**), and kojic acid. Samples shown are oxyresveratrol (circle, ●), 2,4,3',5'-tetrahydroxybibenzyl (triangle, ▲), and kojic acid (diamond, ◆).

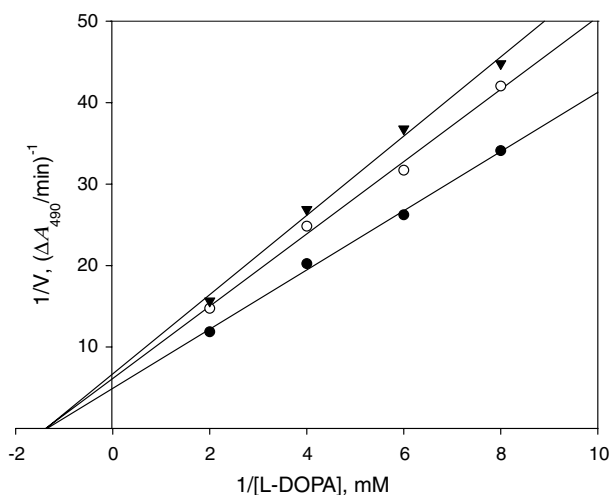
Table 1. Tyrosinase inhibitory activity of **1–8**

Compound	% inhibition (at 100 $\mu$ M)	IC <sub>50</sub> ( $\mu$ M)
2,4,3',5'-Tetrahydroxystilbene ( <b>1</b> )	83.5 $\pm$ 0.9	12.7
<i>trans</i> -2-Methoxy-4,3',5'-trihydroxystilbene ( <b>2</b> )	4.8 $\pm$ 1.6	>100
<i>trans</i> -2,3'-Dimethoxy-4,5'-dihydroxystilbene ( <b>3</b> )	<0 $\pm$ 1.4	>100
<i>trans</i> -4,3'-Dimethoxy-2,5'-dihydroxystilbene ( <b>4</b> )	<0 $\pm$ 3.6	>100
<i>trans</i> -2,4,3',5'-Tetramethoxystilbene ( <b>5</b> )	<0 $\pm$ 2.6	>100
<i>cis</i> -2,4,3',5'-Tetramethoxystilbene ( <b>6</b> )	<0 $\pm$ 1.9	>100
2,4,3',5'-Tetrahydroxybibenzyl ( <b>7</b> )	96.3 $\pm$ 0.6	1.6
2,4,3',5'-Tetramethoxybibenzyl ( <b>8</b> )	2.3 $\pm$ 3.2	>100
Kojic acid	41.1 $\pm$ 1.4	133.4

**Table 2.** Kinetic parameters of mushroom tyrosinase in the presence of **1** or **7**

Inhibitor	Dose ( $\mu\text{M}$ )	$K_m$ (M)	$V_{\max}$ ( $\Delta A_{490}/\text{min}$ )	$K_i$ (M)
None	—	$0.7 \times 10^{-3}$	$2.0 \times 10^{-1}$	—
<b>7</b>	1.2	$0.7 \times 10^{-3}$	$1.6 \times 10^{-1}$	$4.8 \times 10^{-6}$
<b>7</b>	2.4	$0.7 \times 10^{-3}$	$1.4 \times 10^{-1}$	$5.6 \times 10^{-6}$
<b>1</b>	9.2	$0.7 \times 10^{-3}$	$1.4 \times 10^{-1}$	$2.2 \times 10^{-5}$
<b>1</b>	18.4	$0.7 \times 10^{-3}$	$1.2 \times 10^{-1}$	$2.8 \times 10^{-5}$

values with different concentrations of L-dopa (Fig. 3) that the presence of **7** at different concentrations (1.2 and 2.4  $\mu\text{M}$ ) did not affect the  $K_m$  value (0.7 mM L-dopa) of the enzyme, but decreased the  $V_{\max}$  values to  $1.6 \times 10^{-1}$  and  $1.4 \times 10^{-1}$ , respectively (Table 2). Therefore, **7** was a non-competitive inhibitor of tyrosinase on L-dopa with the  $K_i$  values of 4.8–5.6  $\mu\text{M}$  (Table 2). A previous kinetic study on mushroom tyrosinase with L-dopa as the substrate indicated that **1** was a non-competitive inhibitor.<sup>15</sup> This is in agreement with the results



**Figure 3.** Lineweaver–Burk plot of mushroom tyrosinase in the presence of 2,4,3',5'-tetrahydroxybibenzyl (**7**). Data were obtained as mean values of  $1/V$ , inverse of the increase of absorbance at the wavelength 490 nm per min ( $\Delta A_{490}/\text{min}$ ), with different concentrations of L-dopa as substrate. Inhibitors of the enzyme were 2,4,3',5'-tetrahydroxybibenzyl (**7**) with 2.4  $\mu\text{M}$  (triangle,  $\blacktriangledown$ ), 1.2  $\mu\text{M}$  (circle,  $\circ$ ), and no 2,4,3',5'-tetrahydroxybibenzyl (**7**) (circle,  $\bullet$ ).

**Table 3.** Cytotoxicity of **1–8**

Compound	$\text{IC}_{50}$ ( $\mu\text{M}$ )		
	KB	BC	NCI-H187
2,4,3',5'-Tetrahydroxystilbene ( <b>1</b> )	NA <sup>a</sup>	NA	NA
<i>trans</i> -2-Methoxy-4,3',5'-trihydroxystilbene ( <b>2</b> )	NA	66.6	NA
<i>trans</i> -2,3'-Dimethoxy-4,5'-dihydroxystilbene ( <b>3</b> )	5.5	10.8	10.9
<i>trans</i> -4,3'-Dimethoxy-2,5'-dihydroxystilbene ( <b>4</b> )	16.5	13.9	33.5
<i>trans</i> -2,4,3',5'-Tetramethoxystilbene ( <b>5</b> )	8.6	5.6	8.0
<i>cis</i> -2,4,3',5'-Tetramethoxystilbene ( <b>6</b> )	0.3	1.0	0.3
2,4,3',5'-Tetrahydroxybibenzyl ( <b>7</b> )	NA	NA	NA
2,4,3',5'-Tetramethoxybibenzyl ( <b>8</b> )	NA	NA	NA
Ellipticine	2.4	2.7	1.9
Doxorubicin	0.3	0.5	0.1

<sup>a</sup> NA, no activity, showing less than 50% inhibition at 20  $\mu\text{g}/\text{ml}$ .

of our investigation, as reflected in the unchanged  $K_m$  value (0.7 mM L-dopa) in the presence of **1** (Table 2). When the  $K_i$  values of **1** ( $2.2 \times 10^{-5}$ – $2.8 \times 10^{-5}$  M) and **7** ( $4.8 \times 10^{-6}$ – $5.6 \times 10^{-6}$  M) were compared (Table 2), it was found that **7** had about 5-fold higher affinity to the enzyme than **1**. The higher potency of the former is probably due to this kinetic property.

A study of the cytotoxicity of **1–8** was carried out using a battery of cancer cell lines, including KB, BC, and NCI-H187, with the anticancer agents ellipticine and doxorubicin as positive controls.<sup>16,17</sup> Table 3 illustrates the  $\text{IC}_{50}$  values for these compounds. It is obvious that while the parent compound **1** was not cytotoxic, all of its *O*-methylated derivatives (**2–6**) demonstrated cytotoxic potential, ranging from weak to very strong activity. Thus, it appears that the introduction of methoxy groups to the aromatic rings generated cytotoxicity. It should be noted that the *cis*-polymethoxystilbene **6** was more potent than the *trans*-isomer **5**. The bibenzyl structures **7** and **8**, however, were devoid of cytotoxicity. Therefore, it could be deduced from these data that the requirements for cytotoxicity of the stilbenoids were: (1) polymethoxy groups, (2) unsaturation at  $\alpha$ - and  $\beta$ -carbons, and (3) *cis*-configuration. In view of the  $\text{IC}_{50}$  value, **6** showed more cytotoxicity than the positive control ellipticine in all cell lines. However, when compared with doxorubicin, **6** was equally active in KB but less active in BC and NCI-H187 cells. Structurally, stilbene **6** could be considered, as related to combretastatins, a group of natural products possessing strong anticancer potential, which were first isolated from *Combretum caffrum* (Combretaceae).<sup>18</sup> Studies have shown that *trans*-polymethoxystilbenoids exert their cytotoxicity by inhibition of cytochrome P<sub>450</sub> 1B1,<sup>19</sup> whereas *cis*-analogues have inhibitory effects on tubulin polymerization of cancer cells.<sup>18</sup> More studies on **6**, such as in vitro assays in other cancer cell lines and in vivo experiments in animals, as well as examinations of the mechanism of action, are needed before the anticancer potential of this compound can be fully determined.

In summary, this study showed that a more potent tyrosinase inhibitor (**7**) can be obtained from oxyresveratrol (**1**) through a single-step reduction reaction. The bibenzyl structure **7** was a non-competitive inhibitor of mushroom tyrosinase with higher affinity to the enzyme

than **1**, and exhibited no toxicity in human cells such as KB, BC, and NCI-H187. The present study also presented a process for the conversion of the non-cytotoxic stilbene **1** into a strongly cytotoxic compound **6** through two simple chemical reactions. The results from this investigation reflect the importance of natural products chemistry as a tool for finding and developing useful bioactive compounds.

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### Supplementary data

Supplementary data associated with this article can be found, in the online version, at [doi:10.1016/j.bmcl.2006.08.018](https://doi.org/10.1016/j.bmcl.2006.08.018).

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